

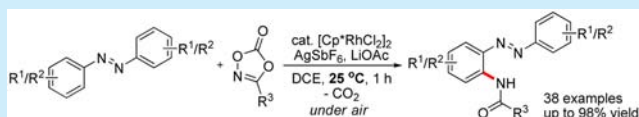
Selective Rhodium-Catalyzed C–H Amidation of Azobenzenes with Dioxazolones under Mild Conditions

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Supporting Information

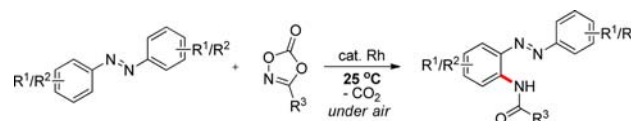
ABSTRACT: A synthetic method for a wide range of amidated azobenzenes is developed from the selective rhodium-catalyzed C–H amidation reaction of symmetrical as well as unsymmetrical azobenzenes with alkyl-, aryl-, and heteroaryl-substituted dioxazolones under mild conditions. Diamidation of azobenzenes and amidation of monoamidated azobenzenes were also demonstrated.



Azobenzenes are a significant class of compounds in many fields, such as organic dyes, food additives, and material science. Thus, development of streamlined synthetic methods for their derivatives is greatly needed.¹ Although the introduction of a wide range of functional groups onto a symmetrical azobenzene ring has been reported, selective introduction of substituents onto unsymmetrical azobenzenes still represents a formidable challenge. In recent years, C–H activation of azobenzenes using azo group as a directing group has gained reasonable attention for the functionalization of azobenzenes including the following reactions: acylation,² alkoxylation,³ nitration,⁴ halogenation,⁵ olefination,⁶ alkylation,⁷ arylation,⁸ phosphorylation,⁹ amination,¹⁰ cyanation,¹¹ amino-carbonylation,¹² and tandem cyclization.¹³ Regardless of the successes that have been achieved with respect to C–H activation of azobenzenes, the formation of a C–N bond through C–H activation of azobenzenes has been scarcely reported. Recently, the Xu, Jia, and Lee groups independently reported Rh-catalyzed C–H amidation of azobenzenes with azides (Scheme 1a, b, and c, respectively).^{10,13d,14} However, these methods are achieved under harsh conditions, and symmetrical azobenzenes are mainly used. In our continuing efforts in C–H activation,¹⁵ we have been interested in the development of direct and selective C–N bond formation using

both symmetrical and unsymmetrical azobenzenes under mild conditions. Chang and co-workers recently reported the usefulness of dioxazolones as excellent amidation reagents in C–H activation.¹⁶ Accordingly, we envisioned the C–H amidation reaction of azobenzenes with dioxazolones. Herein, we demonstrate selective C–H amidation reactions of symmetrical as well as unsymmetrical azobenzenes with a large number of dioxazolones through decarboxylation under mild conditions (Scheme 2).

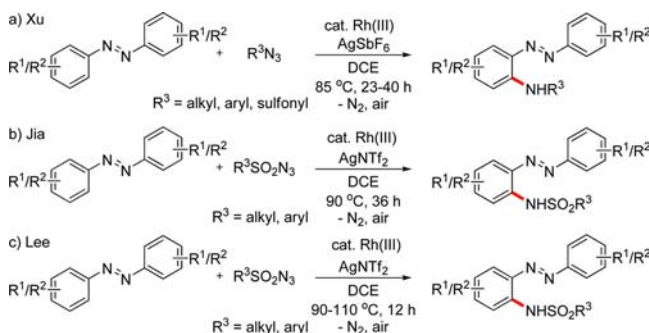
Scheme 2. Selective Rh-Catalyzed C–H Amidation of Azobenzenes with Dioxazolones



First, we began our investigation by examining the reaction of azobenzene (**1a**) with 3-phenyl-1,4,2-dioxazol-5-one (**2a**) in the presence of $[\text{Cp}^*\text{RhCl}_2]_2$ (2.5 mol %) with a variety of silver additives (10.0 mol %), bases (10.0–30.0 mol %), and solvents (Table 1). Among the additives investigated, such as AgNTf_2 , AgOTf , and AgSbF_6 , AgSbF_6 provided the amidated azobenzene **3aa** in 83% yield (entry 3). LiOAc (30.0 mol %) as a base was superior to NaOAc and CsOAc . DCE was the solvent of choice (entry 3); other reaction media such as MeOH, EtOAc, THF, and PhCH_3 were less effective (entries 7–10). Since the desired product **3aa** was obtained in 74% yield without base (entry 6), a reduced amount (10.0 mol %) of LiOAc was used, resulting in the formation of **3aa** in 84% yield (entry 11). The use of $[\text{Cp}^*\text{CoCl}_2]_2$ catalyst was not effective. Azoxybenzene and *N*-methyl-*N*-nitrosoaniline did not react with **2a** under the optimum conditions.

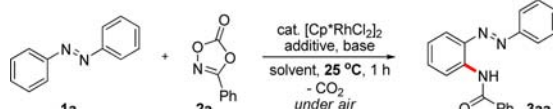
To demonstrate the scope and limitations of the present method, we applied this catalytic system to a wide range of

Scheme 1. Previously Reported Methods for the C–H Amidation of Azobenzenes



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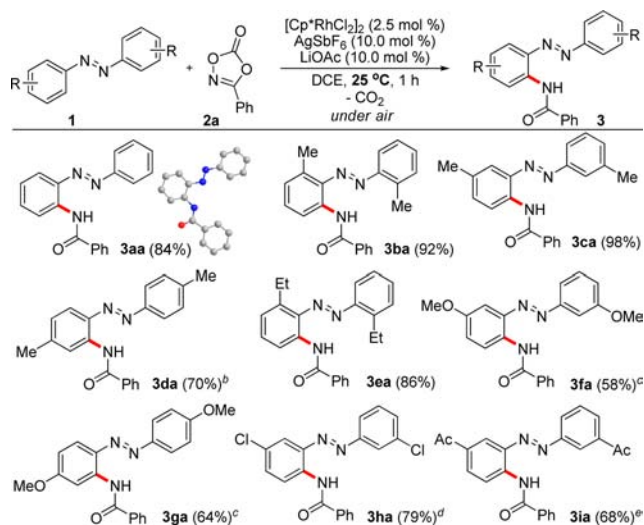
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Table 1. Reaction Optimization^a


entry	additive	base (mol %)	solvent	yield ^b (%)
1	AgNTf ₂	LiOAc (30)	DCE	77
2	AgOTf	LiOAc (30)	DCE	38
3	AgSbF ₆	LiOAc (30)	DCE	83
4	AgSbF ₆	NaOAc (30)	DCE	74
5	AgSbF ₆	CsOAc (30)	DCE	66
6	AgSbF ₆		DCE	74
7	AgSbF ₆	LiOAc (30)	MeOH	0
8	AgSbF ₆	LiOAc (30)	EtOAc	31
9	AgSbF ₆	LiOAc (30)	THF	42
10	AgSbF ₆	LiOAc (30)	PhCH ₃	40
11	AgSbF ₆	LiOAc (10)	DCE	86 (84) ^c

^aConditions: **1a** (2 equiv), **2a** (0.2 mmol, 1 equiv), [Cp*RhCl₂]₂ (2.5 mol %), additive (10.0 mol %), and base (10.0–30.0 mol %) in solvent (1.0 mL) were used under air. ^bNMR yield using dibromomethane as an internal standard. ^cIsolated yield.

symmetrical azobenzenes **1** in reactions with 3-phenyl-1,4,2-dioxazol-5-ones (**2a**) (Scheme 3). When methyl-substituted

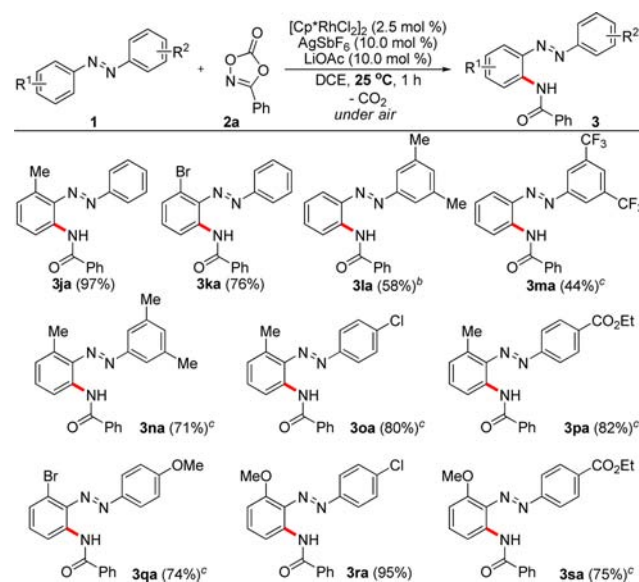
Scheme 3. Substrate Scope of Symmetrical Azobenzenes^a

^aConditions: **1** (2 equiv), **2a** (0.2 mmol, 1 equiv), [Cp*RhCl₂]₂ (2.5 mol %), AgSbF₆ (10.0 mol %), and LiOAc (10.0 mol %) in DCE (1.0 mL) were used at 25 °C for 1 h under air. ^b12 h. ^c[Cp*RhCl₂]₂ (4.0 mol %) and AgSbF₆ (16.0 mol %) were used. ^d**1** (0.2 mmol, 1 equiv) and **2a** (2 equiv) were used for 24 h. ^e60 °C.

azobenzenes were treated with **2a** in the presence of a Rh catalyst at 25 °C, the desired amidation products (**3ba**, **3ca**, and **3da**) were produced in good to excellent yields. In particular, it is noteworthy that 3-methyl-substituted azobenzene **1c** was quantitatively amidated to provide 2-amidated azobenzene **3ca** in 98% yield. In this case, no compound amidated at the 2-position was detected due to steric hindrance. 2-Ethyl-substituted substrate **1e** was also transformed to the amidated azobenzene **3ea** in 86% yield. The strongly electron-donating 3- and 4-methoxy substituents slightly influenced the amidation reaction, and the desired products **3fa** and **3ga** were produced

in 58% and 64% yields, respectively. 3-Chloro-substituted azobenzene **1h** is applicable to the present method, providing the desired azobenzene **3ha** in a good yield. However, 3-acetylazobenzene **1i** was less reactive, and the desired product **3ia** was obtained in 68% yield at 60 °C.

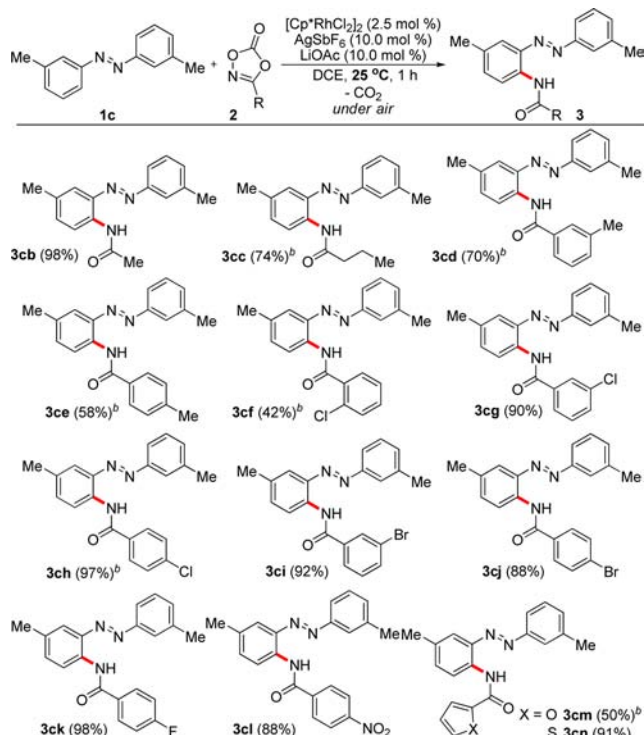
Encouraged by these results, we investigated the selectivity of the Rh-catalyzed C–H amidation reaction using unsymmetrical azobenzenes by modification of the steric and/or electronic environment (Scheme 4). When 2-methylazobenzene **1j** was

Scheme 4. Substrate Scope of Unsymmetrical Azobenzenes^a

^aConditions: **1** (2 equiv), **2a** (0.2 mmol, 1 equiv), [Cp*RhCl₂]₂ (2.5 mol %), AgSbF₆ (10.0 mol %), and LiOAc (10.0 mol %) in DCE (1.0 mL) were used at 25 °C for 1 h under air. ^b60 °C. ^c[Cp*RhCl₂]₂ (4.0 mol %) and AgSbF₆ (16.0 mol %) were used.

treated with the Rh catalyst, C–H amidation reaction selectively took place on the methyl-substituted aryl ring under mild conditions, and the desired amidated azobenzene **3ja** was obtained in 97% yield. In the case of 2-bromoazobenzene **1k**, the C–H amidation reaction also occurred on the bromo-substituted aryl ring. In these cases, no diamidated compounds were detected. These results indicate that the selectivity of the C–H amidation reaction in unsymmetrical azobenzenes is governed by steric effects rather than by electronic effects. Although 3,5-dimethyl- and 3,5-di(trifluoromethyl)azobenzene were less reactive, the amidation reaction selectively occurred on the phenyl ring. 2-Methyl along with 3,5-dimethyl-, 4-chloro-, and 4-(ethoxycarbonyl)-substituted azobenzenes (**1n**, **1o**, and **1p**) selectively underwent a C–H amidation reaction on the phenyl ring having a 2-methyl group. Azobenzene **1q** bearing 2-bromo and 4'-methoxy groups was selectively transformed to 6-amidated azobenzene **3qa** due to steric congestion. 2-Methoxy- along with 4'-chloro- and 4'-(ethoxycarbonyl)azobenzenes (**1r** and **1s**) were subjected to a selective C–H amidation reaction on the phenyl ring having a 2-methoxyl group.

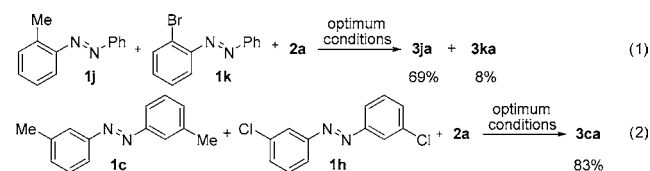
To further investigate the generality of the present method, alkyl-substituted dioxazolones **2** were examined in a reaction with **1c** under the optimum conditions (Scheme 5). C–H amidation reactions using 3-methyl-1,4,2-dioxazol-5-one (**2b**) led to the corresponding product **3cb** in 98% yield. To our

Scheme 5. Substrate Scope of Dioxazolones^a

^aConditions: 1c (2 equiv), 2 (0.2 mmol, 1 equiv), [Cp*RhCl₂]₂ (2.5 mol %), AgSbF₆ (10.0 mol %), and LiOAc (10.0 mol %) in DCE (1.0 mL) were used at 25 °C for 1 h under air. ^b[Cp*RhCl₂]₂ (4.0 mol %) and AgSbF₆ (16.0 mol %) were used.

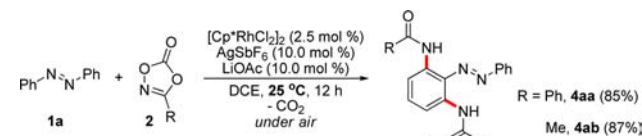
delight, the amidation reaction of 3,3'-dimethylazobenzene (1c) with *n*-propyl-1,4,2-dioxazol-5-one (2c) under the slightly modified reaction conditions (4.0 mol % of Rh catalyst and 16.0 mol % of AgSbF₆) afforded the desired product 3cc in 74% yield. We next focused on the C–H amidation reaction of 1c with a large number of aryl- and heteroaryl-substituted dioxazolones 2. Dioxazolones (2d and 2e) having a methyl group on the phenyl ring were reacted with 1c to give 3cd and 3ce in moderate to good yields. Aryl dioxazolones 2 having electron-withdrawing groups (Cl, Br, F, and NO₂) on the aromatic ring also underwent the amidation reaction to afford the desired products 3cg–cl in good to excellent yields ranging from 88% to 98%. However, 2-chlorophenyl-substituted dioxazolone (2f) was less reactive due to steric effects. Although 3-(2-furyl)-1,4,2-dioxazol-5-one (2m) reacted with 1c to provide 3cm in 50% yield under the modified conditions, 3-(2-thiophenyl)-1,4,2-dioxazol-5-one (2n) gave the desired amidation product 3cn in 91% yield under the optimum conditions.

Competitive reactions between various azobenzenes were performed. Dioxazolone (2a) was simultaneously reacted with 1j and 1k (1 equiv each), which led to the formation of 3ja as the major product (eq 1). Similarly, simultaneous C–H



amidation with 1c and 1h (1 equiv each) resulted in the selective production of 3ca (eq 2). These results indicate that the electron-rich azobenzenes are more reactive than the electron-deficient ones.

Next, we are interested in the diamidation reaction of azobenzenes due to selectivity on the two aryl rings after the first amidation (Scheme 6). When 1a (1 equiv) was treated

Scheme 6. Diamidation of Azobenzenes^a

^aConditions: 1a (0.2 mmol, 1 equiv), 2 (3 equiv), [Cp*RhCl₂]₂ (2.5 mol %), AgSbF₆ (10.0 mol %), and LiOAc (10.0 mol %) in DCE (1.0 mL) were used at 25 °C for 12 h under air.

with 2a (3 equiv) under the optimum conditions, the first as well as second amidation reactions occurred on the same phenyl ring, providing 4aa in 85% yield. 3-Methyl-1,4,2-dioxazol-5-one (2b) turned out to have the same tendency toward a 2-fold Rh-catalyzed diamidation reaction.

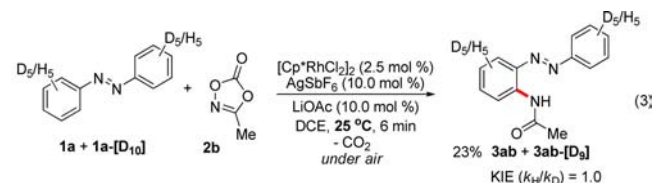
On the basis of these results, the Rh-catalyzed amidation reaction of monoamidated azobenzene 3aa was examined (Scheme 7). When 3aa was treated with methyl- and aryl-

Scheme 7. Amidation of Monoamidated Azobenzenes^a

^aConditions: 3aa (2 equiv), 2 (0.2 mmol, 1 equiv), [Cp*RhCl₂]₂ (2.5 mol %), AgSbF₆ (10.0 mol %), and LiOAc (10.0 mol %) in DCE (1.0 mL) were used at 25 °C for 1 h under air.

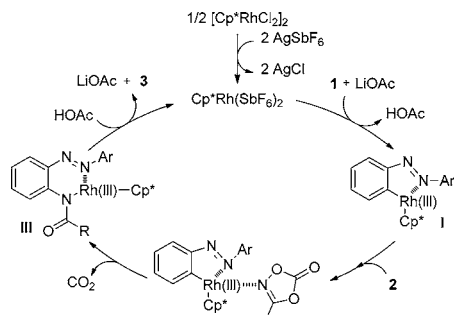
substituted dioxazolones (2b, 2g, 2j, and 2n) under the optimum conditions, the amidation reaction proceeded smoothly, providing the corresponding amidated azobenzenes (5ab–an) in moderate to good yields ranging from 56% to 76%.

Next, we conducted KIE studies to obtain insight into the reaction mechanism (eq 3). A KIE was not observed ($k_H/k_D = 1.0$), indicating that the C–H bond cleavage at the *ortho* position of azobenzene is not involved in the rate-determining step.



A plausible reaction mechanism is described in Scheme 8.¹⁷ First, exposure of [Cp*RhCl₂]₂ to AgSbF₆ and LiOAc provides the cationic Rh(III) catalyst, which undergoes C–H activation with 1 to afford a five-membered rhodacycle I. Subsequent coordination of dioxazolones 2 with rhodacycle I generates intermediate II, which upon decarboxylation and amidation of

Scheme 8. Plausible Mechanism



the C–Rh bond furnishes complex **III**. Finally, protoderhodation of **III** delivers the desired product **3** and the regeneration of the active Rh(III) catalyst.

In summary, we have developed an efficient synthetic method for a wide range of amidated azobenzenes through the selective Rh-catalyzed C–H amidation reaction of symmetrical as well as unsymmetrical azobenzenes with alkyl-, aryl-, and heteroaryl-substituted dioxazolones under mild conditions. In addition, diamidation of azobenzenes and amidation of monoamidated azobenzenes were developed.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b02250.

Experimental procedures, characterization data, X-ray crystallographic data (**3aa**), and NMR spectra for all of the products (PDF)

X-ray data for compound **3aa** (CIF)

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Notes

The authors declare no competing financial interest.

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■ DEDICATION

This paper is dedicated to Professor Naoto Chatani (Osaka University) on the occasion of his 60th birthday.

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